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Key indicators

Single-crystal X-ray study T = 150 K Mean σ (C–C) = 0.008 Å R factor = 0.070 wR factor = 0.208 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Received 6 June 2006

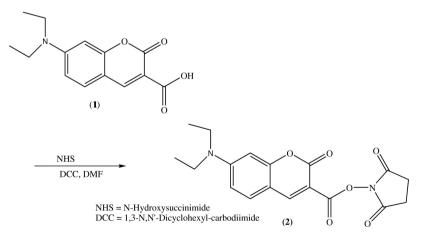
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Succinimidyl 7-(diethylamino)-2-oxo-2*H*chromene-3-carboxylate chloroform solvate

In the title molecular structure, $C_{18}H_{18}N_2O_6$ ·CHCl₃, the dihedral angle between the two fused, essentially planar, sixmembered rings is 5.4 (2)°. In the crystal structure, weak intermolecular C-H···O hydrogen bonds and π - π stacking interactions connect molecules into two-dimensional layers.

Comment

For background information and relevant references see the previous paper (Bardajee et al., 2006). A view of the molecular structure of the title compound, (2), is shown in Fig. 1, and selected bond lengths and angles are given in Table 1. Unlike in the molecular structure of 7-diethylaminocoumarin-3carboxylic acid, (1) (Bardajee et al., 2006), the two fused sixmembered rings in (2) are not coplanar but instead have an angle of 5.4 (2) Å between their least-squares planes. The dihedral angles between the six-membered rings C4-C9 and C1-C4/C9/O1 and the five-membered ring N2/C15-C18 are 73.9 (2) and 77.1 (2) $^{\circ}$, respectively. In the crystal structure, weak C-H···O hydrogen bonds (Table 2) and significant π - π stacking interactions [where $Cg1 \cdots Cg2(\frac{3}{2} - x, \frac{3}{2} - y, 1 - z) =$ 3.695 (3) Å (Cg1 and Cg2 are the centroids defined by ring atoms O1/C1-C4/C9 and C4-C9 respectively) and the perpendicular distance is 3.43 Å] connect molecules into layers, and chloroform solvent molecules are located between these layers (Fig. 2).



Experimental

The title compound was obtained from the reaction of 7-diethylaminocoumarin-3-carboxylic acid, (1), with *N*-hydroxysuccinimide, whose preparation was described previously (Berthelot *et al.*, 2005). In a typical procedure, to a solution of (1) (289.31 mg, 1 mmol) in dimethylformamide (5 ml), *N*-hydroxysuccinimide (115 mg, 1 mmol) was added. After dissolution, the reaction was cooled to 263 K. After

© 2006 International Union of Crystallography All rights reserved stirring for 1 h at this temperature, 1,3-N,N'-dicyclohexylcarbodiimide (226.96 mg, 1.1 mmol) was added. The solution was stirred at 263 K for 12 h. The solution was filtered to remove dicyclohexylurea. A 2-propanol–hexane mixture (1:20) was added to the filtrate to give product (2) as a yellow solid (315 mg, 88%). X-ray quality crystals were obtained by slow recrystallization of (2) from chloroform at 277 K.

Z = 8

 $D_x = 1.526 \text{ Mg m}^{-3}$

 $0.30 \times 0.16 \times 0.08 \; \text{mm}$

11417 measured reflections

3598 independent reflections

2151 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

 $\mu = 0.48 \text{ mm}^{-1}$

T = 150 (1) K

Needle, yellow

 $R_{\rm int} = 0.126$

 $\theta_{\rm max} = 25.2^\circ$

Crystal data

 $\begin{array}{l} C_{18}H_{18}N_2O_6 \cdot CHCl_3 \\ M_r = 477.71 \\ \text{Monoclinic, } C2/c \\ a = 19.601 \ (3) \ \text{\AA} \\ b = 9.3170 \ (14) \ \text{\AA} \\ c = 22.807 \ (3) \ \text{\AA} \\ \beta = 93.429 \ (9)^\circ \\ V = 4157.6 \ (10) \ \text{\AA}^3 \end{array}$

Data collection

 $\begin{array}{l} \mbox{Bruker-Nonius KappaCCD} \\ \mbox{diffractometer} \\ \varphi \mbox{ scans and } \omega \mbox{ scans with } \kappa \mbox{ offsets} \\ \mbox{Absorption correction: multi-scan} \\ (SORTAV; \mbox{ Blessing, 1995}) \\ T_{\rm min} = 0.746, \ T_{\rm max} = 0.964 \end{array}$

Refinement

 $\begin{array}{ll} \mbox{Refinement on } F^2 & w = 1/[\sigma^2(F_o^2) + (0.094P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.070 & + 7.3572P] \\ wR(F^2) = 0.209 & where \ P = (F_o^2 + 2F_c^2)/3 \\ S = 1.05 & (\Delta/\sigma)_{\rm max} < 0.001 \\ 3598 \ {\rm reflections} & \Delta\rho_{\rm max} = 0.30 \ {\rm e} \ {\rm \AA}^{-3} \\ 272 \ {\rm parameters} & \Delta\rho_{\rm min} = -0.39 \ {\rm e} \ {\rm \AA}^{-3} \\ \mbox{H-atom parameters constrained} \\ \end{array}$

Table 1

Selected geometric parameters (Å, °).

1.380 (6)	O4-N2	1.390 (6)
1.393 (7)	O4-C14	1.393 (7)
1.201 (6)	N2-C18	1.383 (8)
1.212 (7)	N2-C15	1.400 (8)
121.4 (5)	C18-N2-C15	115.2 (5)
122.3 (5)	O4-N2-C15	121.4 (5)
116.2 (5)	O2-C1-C2	128.3 (6)
122.2 (5)		
	1.393 (7) 1.201 (6) 1.212 (7) 121.4 (5) 122.3 (5) 116.2 (5)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Table 2

Hydrogen-bond geometry (Å, °).

$\overline{D - \mathbf{H} \cdots A}$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
C1S-H1SA····O6 ⁱ	1.00	2.28	3.240 (7)	160
$C3-H3A\cdots O3^{ii}$	0.95	2.59	3.219 (7)	124
$C5-H5A\cdots O6^{ii}$	0.95	2.50	3.395 (7)	157
$C11 - H11C \cdots O6^{iii}$	0.98	2.53	3.473 (9)	162
$C12-H12B\cdots O3^{iv}$	0.99	2.54	3.358 (8)	140
$C16-H16B\cdots O4^{v}$	0.99	2.42	3.195 (8)	135
	-		. (.)	

Symmetry codes: (i) $x - \frac{1}{2}, y + \frac{1}{2}, z$; (ii) $-x + \frac{3}{2}, -y + \frac{1}{2}, -z + 1$; (iii) $-x + \frac{3}{2}, -y + \frac{3}{2}, -z + 1$; (iv) x, y + 1, z; (v) $-x + \frac{3}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$.

Analysis of the data using *PLATON* (Spek, 2003) revealed that the crystal was a non-merohedral twin and a twin rotation matrix $(-100, 0\overline{10}, 0.139 \ 0 \ 1)$ was applied. In subsequent refinements the

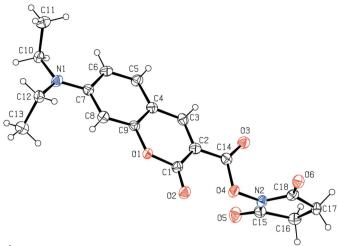


Figure 1

Molecular structure of (2), showing 30% probability displacement ellipsoids (arbitrary spheres for H atoms). The solvent molecule has been omitted.

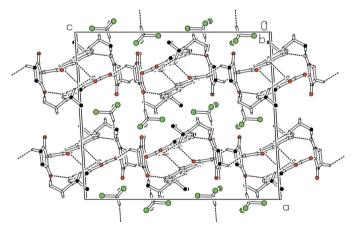


Figure 2

A partial packing plot (Spek, 2003) of (2), showing hydrogen bonds as dashed lines.

value of the weighted *R*-factor (for all data) improved from 0.303 to 0.209. The twin fraction refined to 0.884:0.116 (2). The relatively high $R_{\rm int}$ value is related to the twinning. H atoms bonded to C atoms were placed in calculated positions with C-H = 0.95 Å (aromatic), 0.99 Å (methylene), 0.98 Å (methyl) or 1.00 Å (chloroform) and were included in the refinement in a riding-model approximation with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$, or $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm C})$ for methyl H atoms.

Data collection: *COLLECT* (Nonius, 2002); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXTL* (Sheldrick, 2001); molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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